

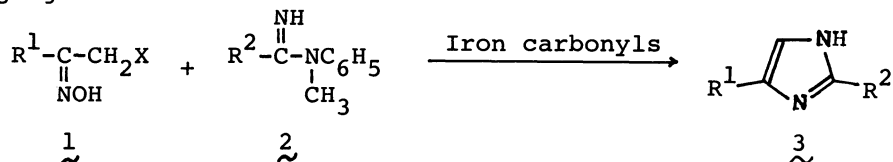
SYNTHESIS OF IMIDAZOLE DERIVATIVES FROM α -HALOOXIMES
AND AMIDINES BY USE OF IRON CARBONYLS

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The reaction of α -halooximes with amidines in the presence of iron carbonyls gives imidazole derivatives in good yields. This reaction occurs via deoxygenation of 4H-1,2,5-oxadiazines by iron carbonyls.

Iron carbonyls induce a variety of organic transformations.¹⁾ They can be utilized even for heterocyclic transformations. For example, azirines can be converted into other heterocycles such as pyrroles,^{2,3)} pyridines³⁾, pyrazines⁴⁾ and isoxazoles,⁴⁾ depending on the structures of azirines and iron carbonyls employed. However, the selectivity in these transformations is not so high that several compounds are usually produced as by-products. Previously, we have shown that 5,6-dihydro-4H-1,2-oxazines, which are accessible from α -bromooximes and enamines, are converted selectively into pyrrole derivatives in high yields by deoxygenation with $\text{Fe}_3(\text{CO})_{12}$.⁵⁾ We now report the synthesis of imidazole derivatives from α -halooximes and amidines by use of iron carbonyls as a deoxygenating agent.



A mixture of α -bromoacetophenone oxime (1a, 1.0 mmol) and N-methyl-N-phenylbenzamidine (2a, 5.0 mmol) in 1,2-dichloroethane (DCE, 20 ml) was stirred at room temperature for 4 h. $\text{Fe}_3(\text{CO})_{12}$ (2.0 mmol) was added and the resulting mixture was heated at 80°C for 20 h. After adding silica gel (5g), the mixture was filtered and the filtrate was evaporated. The residue was chromatographed on silica gel with benzene-ethyl acetate (4:1) to give 2,4-diphenylimidazole (3a) in 74 % yield together with a 64 % recovery of the benzamidine used.

Similar treatment of a mixture of α -halooximes (1a-f) and aromatic amidines (2a-b) with $\text{Fe}_3(\text{CO})_{12}$ yielded the corresponding imidazole derivatives (3a-f). The results are summarized in Table 1.

In these transformations, excess amidines were always needed in order to obtain the imidazoles in high yields although substantially all the unreacted amidines (>80%) were recovered by chromatography from the reaction mixtures.

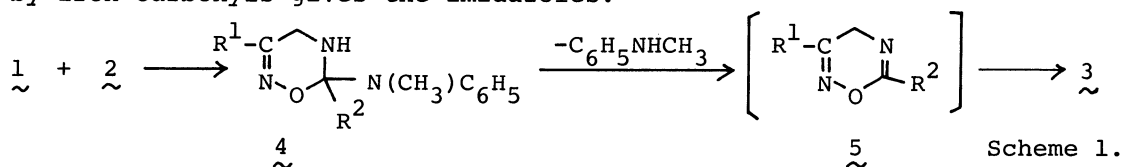
Table 1. Reaction of α -halooximes with amidines in the presence of $\text{Fe}_3(\text{CO})_{12}$

$\underline{1}$	Oxime	X	$\underline{2}$	Amidine	$\underline{3}$	Yield, % ^{a)}
	R^1			R^2		
$\underline{1a}$	C_6H_5	Br	$\underline{2a}$	C_6H_5	$\underline{3a}$	74
	C_6H_5	Br	$\underline{2b}$	<i>m</i> - $\text{CH}_3\text{C}_6\text{H}_4$	$\underline{3b}$	79
$\underline{1b}$	C_6H_5	Cl	$\underline{2a}$	C_6H_5	$\underline{3a}$	68
$\underline{1c}$	<i>p</i> - $\text{CH}_3\text{C}_6\text{H}_4$	Br	$\underline{2a}$	C_6H_5	$\underline{3c}$	69
$\underline{1d}$	<i>p</i> - BrC_6H_4	Br	$\underline{2a}$	C_6H_5	$\underline{3d}$	73
$\underline{1e}$	CH_3	Cl	$\underline{2a}$	C_6H_5	$\underline{3e}$	59
$\underline{1f}$	$\text{C}_2\text{H}_5\text{OCO}$	Br	$\underline{2a}$	C_6H_5	$\underline{3f}$	31

a) Isolated yields based on oximes used.

No imidazole derivatives were obtained from aliphatic amidines such as *N*-methyl-*N*-phenylacetamidine and *N*-methyl-*N*-phenylbutamidine. Among iron carbonyls examined, $\text{Fe}_3(\text{CO})_{12}$ was most effective for the above transformations; the efficiency of iron carbonyls for these transformations decreased in the following order: $\text{Fe}_3(\text{CO})_{12} > \text{Fe}_2(\text{CO})_9 > \text{Fe}(\text{CO})_5$.⁶⁾

A possible pathway of the reaction is shown in Scheme 1. α -Halooximes react with amidines to give 5,6-dihydro-4*H*-1,2,5-oxadiazines ($\underline{4}$) by elimination of hydrogen halide.⁷⁾ A base-catalyzed elimination of *N*-methylaniline from $\underline{4}$ affords the corresponding 4*H*-1,2,5-oxadiazines ($\underline{5}$).⁸⁾ Finally, the deoxygenation of $\underline{5}$ by iron carbonyls gives the imidazoles.



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- The reactions of $\underline{1a}$ (1.0 mmol) and $\underline{2a}$ (5.0 mmol) were carried out in the presence of $\text{Fe}(\text{CO})_5$ (6.0 mmol), $\text{Fe}_2(\text{CO})_9$ (3.0 mmol) and $\text{Fe}_3(\text{CO})_{12}$ (2.0 mmol) under similar conditions described in text, giving $\underline{3a}$ in the yields of 54 % for $\text{Fe}(\text{CO})_5$, 60 % for $\text{Fe}_2(\text{CO})_9$ and 74 % for $\text{Fe}_3(\text{CO})_{12}$.
- Treatment of $\underline{1a}$ with $\underline{2a}$ in DCE at room temperature without added $\text{Fe}_3(\text{CO})_{12}$ gave 5,6-dihydro-3,6-diphenyl-4*H*-6-(*N*-methylanilino)-1,2,5-oxadiazine $\underline{4a}$ in 51 % yield, mp 134-135°C :IR(KBr) 3280 cm^{-1} (νNH); NMR(CDCl_3) δ =3.10(3H,s, NCH_3) 5.19(2H,d, J =6 Hz, CH_2NH), 6.5-6.7(1H,m,NH), 6.9-7.9 ppm(15H,m, C_6H_5). Compound $\underline{4a}$ was converted to $\underline{3a}$ by treating with $\text{Fe}_3(\text{CO})_{12}$ in 93 % yield.
- N*-Methylaniline was obtained in high yield in all the imidazole-forming reactions.

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